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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=294 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.010 \AA$
Disorder in main residue
$R$ factor $=0.067$
$w R$ factor $=0.156$
Data-to-parameter ratio $=12.4$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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## 1,2-Bis[5-(2,4-difluorophenyl)-2-methyl-3-thienyl]-3,3,4,4,5,5-hexafluorocyclopent-1-ene: a new photochromic diarylethene compound

The title compound, $\mathrm{C}_{27} \mathrm{H}_{14} \mathrm{~F}_{10} \mathrm{~S}_{2}$, is a new symmetrical photochromic diarylethene derivative which can be used potentially for optical recording media and other optoelectronic device materials. There are two independent molecules in the asymmetric unit. Both adopt a photoactive antiparallel conformation. The distances between the reactive C atoms in the two molecules are 3.664 (7) and 3.621 (7) A. For one molecule, the dihedral angles between the central cyclopentene ring and the thiophene rings are 46.4 (9) and $51.9(9)^{\circ}$, and those between the thiophene rings and the adjacent benzene rings are 2.1 (8) and $18.9(8)^{\circ}$. For the other molecule, the dihedral angles between the central cyclopentene ring and the thiophene rings are 134.5 (9) and $44.3(9)^{\circ}$, and those between the thiophene rings and the adjacent benzene rings are 2.7 (8) and 27.7 (8) ${ }^{\circ}$.

## Comment

Photochromic diarylethene crystals are very interesting not only in designing new materials for optical data storage, but also because the photoinduced molecular transformations might be used to gain control over other physical properties in the solid state (Kobatake \& Irie, 2004; Chai et al., 2005). To date, many diarylethenes have been reported (Irie, 2000; Matsuda \& Irie, 2004; Tian \& Yang, 2004; Pu, Li et al., 2006; Pu, Yang et al., 2006; Yang et al., 2006). For further background information, see Pu, Yang et al. (2005). The present paper presents the crystal structure of the title compound, (1a).

(3)

(1a)

The molecular structure of ( $1 a$ ) is shown in Fig. 1. Relevant bond lengths and torsion angles are given in Table 1. There are

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Figure 1
The structures of the two independent molecules in the asymmetric unit of ( $1 a$ ), with $35 \%$ probability displacement ellipsoids, showing the atomic numbering scheme. The minor components of disorder for F atoms are not shown.
two molecules in the asymmetric unit and both of them have a photoactive antiparallel conformation. The F atoms (F9 and F19) are disordered over one ortho position of one benzene ring of each molecule (Fig. 1). In the central cyclopentene ring of the two molecules, the $\mathrm{C} 12=\mathrm{C} 16$ and $\mathrm{C} 39=\mathrm{C} 43$ bonds are clearly double bonds and the other bonds to atoms C12, C16, C39 and C43 are clearly single bonds (Table 1).

For one molecule, the two methyl groups are located on different sides of the $\mathrm{C} 12=\mathrm{C} 16$ bond, reflected in the torsion angles $\mathrm{C} 10-\mathrm{C} 9-\mathrm{C} 12=\mathrm{C} 16$ and $\mathrm{C} 12=\mathrm{C} 16-\mathrm{C} 17-\mathrm{C} 18$, and are thus trans with respect to the double bond. Such a conformation is crucial to its photochromic and photoinduced properties (Woodward \& Hoffmann, 1970). The two independent planar thiophene ring systems have essentially identical geometry, and the dihedral angles between the central cyclopentene ring and the thiophene rings are 46.4 (9) ${ }^{\circ}$ for $\mathrm{S} 1 /$ C7-C10 and $51.9(9)^{\circ}$ for S2/C18/C17/C20/C21, while those between the thiophene rings and the adjacent benzene rings are $2.1(8)^{\circ}$ for $\mathrm{C} 1-\mathrm{C} 6$ and $18.9(8)^{\circ}$ for $\mathrm{C} 22-\mathrm{C} 27$. For the other molecule, the two methyl groups are also located on different sides of $\mathrm{C} 12=\mathrm{C} 16$ bond, reflected in the torsion angles $\mathrm{C} 37-\mathrm{C} 36-\mathrm{C} 39=\mathrm{C} 43$ and $\mathrm{C} 39=\mathrm{C} 43-\mathrm{C} 44-\mathrm{C} 45$. The dihedral angles between the central cyclopentene ring and the thiophene rings are $134.5(9)^{\circ}$ for S3/C34-C37 and 44.3 (9) ${ }^{\circ}$ for $\mathrm{S} 4 / \mathrm{C} 45 / \mathrm{C} 44 / \mathrm{C} 47 / \mathrm{C} 48$, and those between the thiophene rings and the adjacent benzene rings are 2.7 (8) ${ }^{\circ}$ for C28-C33 and 27.7 (8) ${ }^{\circ}$ for C49-C54. The distances between the reactive C atoms $(\mathrm{C} 10 \cdots \mathrm{C} 18$ and $\mathrm{C} 37 \cdots \mathrm{C} 45)$ in the two


Figure 2
A packing diagram for ( $\mathrm{I} a$ ), viewed down the $a$ axis. Hydrogen bonds are shown as dashed lines.
molecules are 3.664 (7) and 3.621 (7) $\AA$. This distance demonstrates that this crystal can undergo photochromism in the crystalline phase (Ramamurthy \& Venkatesan, 1987; Kobatake et al., 2004).

## Experimental

Compound (1a) was derived originally from 2-methylthiophene. First, 3-bromo-2-methyl-5-(2,4-difluorophenyl)thiophene, (3) ( 2.65 g , 9.1 mmol ), was prepared in $70 \%$ yield by reacting 2-bromo-2-methyl-5-thienylboronic acid (2) ( $2.87 \mathrm{~g}, 13 \mathrm{mmol}$ ) (Miyasaka et al., 1997; Pu, Liu et al., 2005) with 1,5-difluoro-2-bromobenzene ( $2.51 \mathrm{~g}, 13 \mathrm{mmol}$ ) in the presence of tetrakis(triphenylphosphine)palladium(0) $(0.5 \mathrm{~g})$ and $\mathrm{Na}_{2} \mathrm{CO}_{3}(5.3 \mathrm{~g}, 50 \mathrm{mmol})$ in THF ( 80 ml containing $10 \%$ water), for 16 h at 343 K . Under an argon gas atmosphere, compound (3) ( $2.31 \mathrm{~g}, 8 \mathrm{mmol}$ ) was dissolved in THF ( 35 ml ) and $n$-butyl lithium hexane solution ( 3.2 ml of $2.5 \mathrm{~mol} \mathrm{l}^{-1}$ ) was added at 273 K . Stirring was continued for 30 min at this low temperature, and then octafluorocyclopentene ( $0.55 \mathrm{ml}, 4 \mathrm{mmol}$ ) was added and stirred for 2 h at this temperature. After extracting with diethyl ether and evaporation in vacuo, the residue was purified by column chromatography on silica gel (hexane) to give 0.95 g of the title compound in $40 \%$ yield. Crystals suitable for X-ray analysis were grown from a chloroform solution by slow evaporation at room temperature (m.p. 374 K ).

## Crystal data

## $\mathrm{C}_{27} \mathrm{H}_{14} \mathrm{~F}_{10} \mathrm{~S}_{2}$

$M_{r}=592.50$
Triclinic, $P \overline{1}$
$a=11.873$ (5) Å
$b=13.106$ (5) $\AA$
$c=16.653$ (7) $\AA$
$\alpha=101.505(8)^{\circ}$
$\beta=94.504(9)^{\circ}$
$\gamma=92.914(8)^{\circ}$

$$
\begin{aligned}
& V=2525.6(18) \AA^{3} \\
& Z=4 \\
& D_{x}=1.558 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \mu=0.30 \mathrm{~mm}^{-1} \\
& T=294(2) \mathrm{K} \\
& \text { Prism, colorless } \\
& 0.24 \times 0.22 \times 0.12 \mathrm{~mm}
\end{aligned}
$$

## Data collection

## Bruker SMART CCD area-detector

 diffractometer$\varphi$ and $\omega$ scans
Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $T_{\text {min }}=0.757, T_{\text {max }}=1.000$

> 12920 measured reflections 8867 independent reflections 3277 reflections with $I>2 \sigma(I)$ $R_{\text {int }}=0.051$ $\theta_{\max }=25.0^{\circ}$

## Refinement

| Refinement on $F^{2}$ | $w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0152 P)^{2}\right.$ |
| :--- | :---: |
| $R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.067$ | $+5 P]$ |
| $w R\left(F^{2}\right)=0.156$ | where $P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3$ |
| $S=1.01$ | $(\Delta / \sigma)_{\max }<0.001$ |
| 8867 reflections | $\Delta \rho_{\max }=0.75 \mathrm{e} \AA^{-3}$ |
| 713 parameters | $\Delta \rho_{\min }=-0.29 \mathrm{e} \mathrm{A}^{-3}$ |
| H-atom parameters constrained |  |

Table 1
Selected geometric parameters $\left(\AA{ }^{\circ}{ }^{\circ}\right)$.

| S1-C10 | $1.717(7)$ | C12-C13 | $1.500(8)$ |
| :--- | :---: | :--- | :---: |
| S1-C7 | $1.729(6)$ | C13-C14 | $1.532(9)$ |
| S2-C18 | $1.725(6)$ | C14-C15 | $1.520(9)$ |
| S2-C21 | $1.737(6)$ | C15-C16 | $1.501(8)$ |
| S3-CC37 | $1.707(7)$ | C39-C43 | $1.355(8)$ |
| S3-C34 | $1.732(6)$ | C39-C40 | $1.524(8)$ |
| S4-C45 | $1.713(7)$ | C40-C41 | $1.523(9)$ |
| S4-C48 | $1.734(7)$ | C41-C42 | $1.540(9)$ |
| C12-C16 | $1.345(8)$ | C42-C43 | $1.508(8)$ |
|  |  |  |  |
| C12-C9-C10-S1 | $-177.6(5)$ | C12-C16-C17-C18 | $-53.3(10)$ |
| C10-C9-C12-C16 | $-46.3(11)$ | C37-C36-C39-C43 | $-47.9(11)$ |
| C10-C9-C12-C13 | $135.1(7)$ | C35-C36-C39-C40 | $-51.2(9)$ |
| C8-C9-C12-C13 | $-41.9(9)$ | C40-C41-C42-C43 | $-27.1(7)$ |
| C16-C12-C13-C14 | $10.6(8)$ | C40-C39-C43-C44 | $176.6(6)$ |
| C13-C14-C15-C16 | $9.8(8)$ | C41-C42-C43-C39 | $22.0(7)$ |
| C13-C12-C16-C15 | $-4.4(8)$ | C39-C43-C44-C45 | $-42.4(11)$ |
| C14-C15-C16-C12 | $-3.7(8)$ |  |  |

The F atoms at C23 and C54 were found to be disordered over two distinct conformations. From refinement (anisotropic), the site occupancies were fixed at 0.70:0.30 for these F atoms. All H atoms were placed in calculated positions, with $\mathrm{C}-\mathrm{H}$ distances of $0.93 \AA$ (aromatic) and $0.96 \AA\left(\mathrm{CH}_{3}\right)$. They were included in the refinement in the riding-model approximation with isotropic displacement parameters set equal to $1.2 U_{\text {eq }}$ of the carrier atom for the aromatic H and $1.5 U_{\text {eq }}$ of the carrier for $\mathrm{CH}_{3}$.

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

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